

Curriculum Map Year 11 Combined Science - Chemistry

Topic Name	Term	Skills developed with link to NC Subject content	Reflection on previous link in the curriculum	Progress to future link in the curriculum
Organic Chemistry	Autumn HT1	<ul style="list-style-type: none"> Crude oil, hydrocarbons and alkanes Fractional distillation and petrochemicals Properties of hydrocarbons Cracking and alkenes 	<p>Year 7: Separating Mixtures</p> <ul style="list-style-type: none"> condensation evaporation Simple distillation of inky water <p>GCSE Covalent bonding and properties of simple covalent molecules</p> <ul style="list-style-type: none"> This content allows pupils to make the link between molecule size and boiling point due to relative strength of intermolecular forces. 	<p>A level: Organic Chemistry</p> <ul style="list-style-type: none"> Nomenclature Synthesis Reaction conditions Biochemistry Esters Acylation Polymers
Rate of Reaction and Extent of Chemical Change	Autumn HT2	<ul style="list-style-type: none"> Calculating rates of reactions Factors which affect the rates of chemical reactions Collision theory and activation energy Catalysts Reversible reactions Energy changes and reversible reactions Equilibrium The effect of changing conditions on equilibrium (HT only) The effect of changing concentration (HT only) The effect of temperature changes on equilibrium (HT only) The effect of pressure changes on equilibrium (HT only) 	<p>Year 7 Rates of Reaction practical</p> <ul style="list-style-type: none"> Students investigate the surface area of Jelly babies and time taken to dissolve in water. A graph is plotted Particle theory and collisions between water and Jelly baby particles are discussed. 	<p>A Level</p> <ul style="list-style-type: none"> Kinetics and Rate Equations Le Chatelier's Principle Maxwell Boltzmann Distributions Reaction Orders
Chemical Analysis	Spring HT3	<ul style="list-style-type: none"> Pure substances Formulations Chromatography Identification of common gases 	<p>Year 7 Solubility Chromatography</p> <ul style="list-style-type: none"> Simple paper chromatography of pen ink introduces students to R_f values and the relationship between solubility and distance travelled in solvent. <p>Year 8 Reactions of Metals with acids and metal carbonates with acid</p> <ul style="list-style-type: none"> Test for hydrogen Test for carbon dioxide 	<p>KS5 Chemical Analysis</p> <ul style="list-style-type: none"> NMR IR TOF Mass Spectrometry Ion Tests GC Flame Emission Column Chromatography TLC <p>Group 2</p> <ul style="list-style-type: none"> Barium Chloride test for sulfate <p>Group 7</p> <ul style="list-style-type: none"> Testing for halide ions

Atmospheric Chemistry and Using Resources	<i>Spring HT4</i>	<ul style="list-style-type: none"> • The proportions of different gases in the atmosphere • The Earth's early atmosphere • How oxygen increased • How carbon dioxide decreased • Greenhouse gases • Human activities which contribute to an increase in greenhouse gases in the atmosphere • Global climate change • The carbon footprint and its reduction • Atmospheric pollutants from fuels • Properties and effects of atmospheric pollutants • Using the Earth's resources and sustainable development • Potable water • wastewater treatment • Alternative methods of extracting metals (HT only) • Life cycle assessment • Ways of reducing the use of resources 	<p>Year 8: Fuels</p> <ul style="list-style-type: none"> • complete and incomplete combustion <p>Year 8: Climate Change</p> <ul style="list-style-type: none"> • global warming • acid rain <p>Year 8: Sustainability</p> <ul style="list-style-type: none"> • Recycling • Life Cycle assessments • Impact of metal extraction 	<p>A Level</p> <ul style="list-style-type: none"> • Free Radical Chemistry • Atmospheric Pollutants • Polymer disposal • Choosing Suitable Reagents <p>Use of alternative fuels such as Biodiesel from lipids.</p>
Preparation for Examinations	<i>Summer HT5</i>			